

## **Theoretical Investigation of Explosive Detection via Halogen Bonding**

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There are many kinds of widely available and powerful explosives that are extremely hard to detect due to their low-volatility and non-aromatic structure. A more effective explosive detection method would be extremely valuable in protecting the general public. This project studied the possibility of using halogen bonding as the basis for a molecular recognition device for explosives. A halogen bond is a highly directional non-covalent interaction between a region of positive electrostatic potential on a halogen atom and a Lewis base. The interactions of various halogen systems and explosives, which acted as the Lewis bases, were modeled with density functional theory and a multitude of functionals and basis sets in order to ensure accurate energy calculations. The data collected supports the concept that halogen bonding can be used as the basis for a molecular recognition device for explosives.