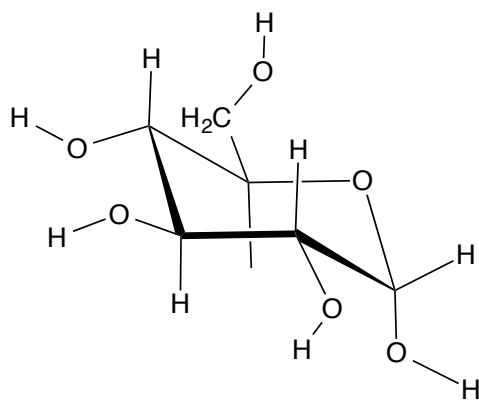


# A Computational Evaluation of the Structures of $\alpha$ - and $\beta$ -D-Glucopyranose and the Contributions to their Stability.

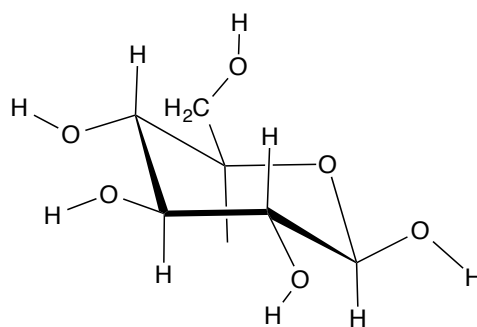
Maryam Abdulsalam, Christian Brutofsky and Marc L. Kasner

Department of Chemistry and Biochemistry  
Montclair State University, Montclair, New Jersey

Geometry optimizations at a variety of computational levels were carried out on  $\alpha$ - and  $\beta$ -D-glucopyranose and a number of model compounds in order to separate the steric and stereoelectronic contributions to the energies and relative stabilities of the  $\alpha$ - versus  $\beta$ - conformers and the *gg*, *gt*, and *tg* rotamers. Changes in the geometry of the conformers give some indicators of the magnitude of the stereoelectronic contributions. Relaxed scans of several dihedrals at the B3LYP/6-311G++(d,p) level show that only the hydroxymethyl substituent on the pyran ring makes any contribution to the set of possible configurations. The relative energies of *gg*, *gt*, and *tg* rotamers from these scans are consistent with other computational studies at this and other levels of calculations. Comparisons of dihedral scans and conformational energies of 2-hydroxymethyl 6-hydroxyl-tetrahydro-2*H*-pyran and 2-hydroxymethyl-tetrahydro-2*H*-pyran with 1,5-anhydro-D-glucitol and D-glucopyranose indicate that the hydroxyl substituents on the pyran ring make little contribution to the stability of the  $\alpha$ -,  $\beta$ - conformers and the *gg*, *gt*, and *tg* rotamers of D-glucopyranose.



(1a)  $\alpha$ -Glucopyranose



(1b)  $\beta$ -Glucopyranose